

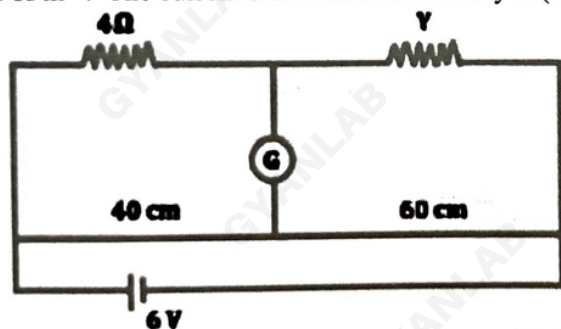
**MHT CET – 2021**  
**23<sup>rd</sup> September (Shift - 1)**

**Section I**

**PHYSICS**

1. If  $m'$  represents the mass of each molecules of a gas and  $T'$  its absolute temperature then the root mean square speed of the gas molecule is proportional to  
(A)  $m \frac{1}{2} T^2$       (B)  $mT$       (C)  $m^2 T^{\frac{1}{2}}$       (D)  $m^2 T^{\frac{1}{2}}$
2. The mass of a spherical planet is 4 times the mass of the earth, but its radius ( $R$ ) is same as that of the earth. How much work is done in lifting a body of mass 5 kg through a distance of 2 m on the planet? ( $g = 10 \text{ ms}^{-2}$ )  
(A) 400 J      (B) 200 J      (C) 800 J      (D) 300 J
3. For series LCR circuit, which one of the following is a CORRECT statement?  
(A) Potential difference across resistance  $R$  and that across capacitor have phase difference  $\frac{\pi^c}{2}$ .  
(B) Applied e.m.f. and potential difference across resistance ' $R$ ' are in the same phase  
(C) Applied e.m.f. and potential difference inductor coil has phase difference of  $\frac{\pi^c}{2}$   
(D) Potential difference across capacitor and that across inductor have phase difference of  $\frac{\pi^c}{2}$ .
4. Two waves  $Y_1 = 0.25 \sin 316t$  and  $Y_2 = 0.25 \sin 310 t$  are propagation same direction. The number of beats produced per second are  
(A)  $\frac{3}{\pi}$       (B)  $\frac{\pi}{3}$       (C)  $\frac{\pi}{2}$       (D)  $\frac{2}{\pi}$
5. In an LCR series a.c. circuit, the voltage across each of the components  $L$ ,  $C$  and  $R$  is 60 V. The voltage across the LC combination is  
(A) 120 V      (B) 60 V      (C) zero V      (D)  $\frac{60}{\sqrt{3}}$  V
6. The speed of a ball of radius 2 cm in a viscous liquid is 20 cm/s. What will be the speed of a ball of radius 1 cm in same liquid?  
(A) 10 cm/s      (B) 4 cm/s      (C) 5 cm/s      (D) 8 cm/s
7. In hydrogen atom an electron revolves around a proton (in nucleus) at a distance ' $r$ ' m. the intensity of electric field due to the proton at distance ' $r$ ' is  $5 \times 10^{11} \text{ NC}^{-1}$ , the magnitude of force between the electron and proton is [charge on electron =  $1.6 \times 10^{-19} \text{ C}$ ]  
(A)  $4 \times 10^8 \text{ N}$       (B)  $8 \times 10^8 \text{ N}$       (C)  $4 \times 10^{-8} \text{ N}$       (D)  $8 \times 10^{-8} \text{ N}$
8. A particle performing S.H.M. when displacement is ' $x$ ', the potential energy and restoring force acting on it are denoted by ' $E$ ' and ' $F$ ' respectively. The relation between  $x$ ,  $E$  and  $F$  is  
(A)  $\frac{2E}{F} - x^2 = 0$       (B)  $\frac{2E}{F} + x^2 = 0$       (C)  $\frac{2E}{F} + x = 0$       (D)  $\frac{2E}{F} - x = 0$

9. A molecule consists of two atoms each of mass 'm' and separated by a distance 'd'. At room temperature, if the average rotational kinetic energy is 'E' then the angular frequency is  
 (A)  $\frac{2}{d} \sqrt{\frac{E}{m}}$  (B)  $\frac{d}{2} \sqrt{\frac{m}{E}}$  (C)  $\sqrt{\frac{Ed}{m}}$  (D)  $\sqrt{\frac{m}{Ed}}$
10. The magnifying power of a refracting type of astronomical telescope is 'm'. If focal length of eyepiece is doubled then the magnifying power will become  
 (A) m (B) 2m (C)  $\frac{m}{2}$  (D)  $\frac{m}{4}$
11. A wooden block of mass 'm' moves with velocity 'V' and collides with another block of mass '4m', which is at rest. After collision the block of mass 'm' comes to rest. The coefficient of restitution will be  
 (A) 0.7 (B) 0.25 (C) 0.4 (D) 0.5
12. Two charged metallic spheres are joined by a very thin metal wire. If the radius of the larger sphere is four times that of the smaller sphere, the electric field near the larger sphere is  
 (A) twice that near the smaller sphere (B) quarter of that near the smaller sphere  
 (C) same as that near the smaller sphere (D) half of that near smaller sphere
13. Water rises to a height of 2 cm in a capillary tube. If cross-sectional area of the tube is reduced to  $\frac{1}{16}$ <sup>th</sup> of initial area then water will rise to a height of  
 (A) 4 cm (B) 8 cm (C) 12 cm (D) 16 cm
14. The radius of a planet is twice the radius of the earth. Both have almost equal average mass densities. If ' $V_P$ ' and ' $V_E$ ' are escape velocities of the planet and the earth respectively, then  
 (A)  $V_E = 1.5 V_P$  (B)  $V_P = 1.5 V_E$  (C)  $V_P = 2 V_E$  (D)  $V_E = 3 V_P$
15. The shortest wavelength for Lyman series is 912 Å. The longest wavelength in Paschen series is  
 (A) 1216 Å (B) 3646 Å (C) 18760 Å (D) 8208 Å
16. A balanced bridge is shown in the circuit diagram. The metre bridge wire has resistance  $1 \Omega m^{-1}$ . The current drawn from the battery is (Internal resistance of battery is negligible)



- (A) 0.44 A (B) 0.66 A (C) 0.88 A (D) 0.22 A
17. In Young's experiment with a monochromatic source and two slits, one of the slits is covered with black opaque paper, the fringes will  
 (A) be darker (B) be narrower (C) be broader (D) not be observed



18. A ray of light travels from air to water to glass and again from glass to air. Refractive index of water w.r.t. air is 'X', glass w.r.t. water is 'Y' and air w.r.t. glass is 'Z'. Which one of the following is correct?  
(A)  $YZ = X$  (B)  $XYZ = 1$  (C)  $XY = Z$  (D)  $XZ = Y$

19. In the Bohr model, an electron moves in a circular orbit around the nucleus. Considering an orbiting electron to be a circular current loop, the magnetic moment of the hydrogen atom, when the electron is in  $n$ th excited state, is  
( $e$  = electronic charge,  $m_e$  = mass of the electron,  $h$  = Planck's constant)  
(A)  $\left(\frac{e}{m_e}\right)\frac{nh}{2\pi}$  (B)  $\left(\frac{e}{m_e}\right)\frac{n^2h}{2\pi}$  (C)  $\left(\frac{e}{2m_e}\right)\frac{n^2h}{2\pi}$  (D)  $\left(\frac{e}{2m_e}\right)\frac{nh}{2\pi}$

20. In potentiometer experiment, cells of e.m.f. ' $E_1$ ' and ' $E_2$ ' are connected in series ( $E_1 > E_2$ ) the balancing length is 64 cm of the wire. If the polarity of  $E_2$  is reversed, the balancing length becomes 32 cm. The ratio  $\frac{E_1}{E_2}$  is  
(A) 1 : 1 (B) 6 : 1 (C) 3 : 1 (D) 2 : 1

21. Three solid spheres each of mass ' $M$ ' and radius ' $R$ ' are arranged as shown in the figure. The moment of inertia of the system about  $YY'$  will be



- (A)  $\frac{16}{5}MR^2$  (B)  $\frac{21}{5}MR^2$   
(C)  $\frac{7}{5}MR^2$  (D)  $\frac{11}{5}MR^2$

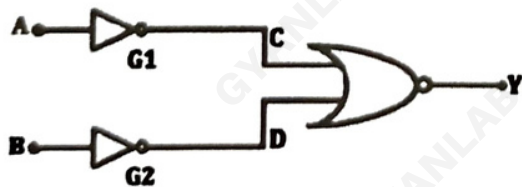
22. A body is performing S.H.M. of amplitude ' $A$ '. The displacement of the body from a point where kinetic energy is maximum to a point where potential energy is maximum, is

- (A) zero (B)  $\pm A$  (C)  $\pm \frac{A}{2}$  (D)  $\pm \frac{A}{4}$

23. Eddy currents are produced when

- (A) a thick metal plate is kept in a steady magnetic field  
(B) a circular coil is placed in a steady magnetic field  
(C) a steady current is passed through a coil  
(D) a thick metal plate is kept in a varying magnetic field

24. The resultant gate and its Boolean expression in the given circuit is



- (A) NOR,  $\overline{A+B}$  (B) AND,  $A \cdot B$  (C) OR,  $A+B$  (D) NAND,  $\overline{A \cdot B}$

25. When wavelength of incident radiation on the metal surface is reduced from ' $\lambda_1$ ' to ' $\lambda_2$ ', the kinetic energy of emitted photoelectrons is tripled. The work function of metal  
[ $h$  = Planck's constant,  $c$  = velocity of light]

- (A)  $\frac{hc}{2} \left[ \frac{3\lambda_1 - \lambda_2}{\lambda_1\lambda_2} \right]$  (B)  $\frac{hc}{2} \left[ \frac{3\lambda_2 - \lambda_1}{\lambda_1\lambda_2} \right]$  (C)  $hc \left[ \frac{3\lambda_1 - \lambda_2}{\lambda_1\lambda_2} \right]$  (D)  $hc \left[ \frac{3\lambda_2 - \lambda_1}{\lambda_1\lambda_2} \right]$

26. A student is throwing balls vertically upwards such that he throws the 2<sup>nd</sup> ball when the 1<sup>st</sup> ball reaches maximum height. If he throws balls at an interval of 3 second, the maximum height of the balls is ( $g = 10 \frac{\text{m}}{\text{s}^2}$ )

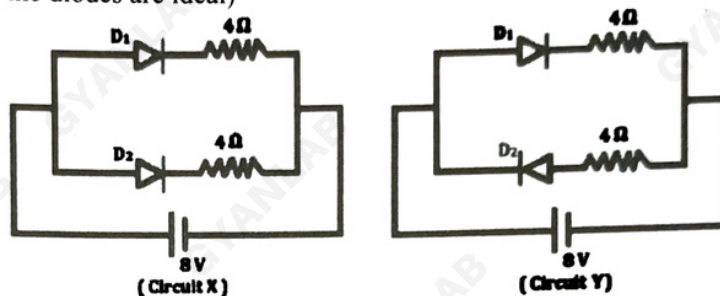
- (A) 45 m (B) 35 m (C) 25 m (D) 30 m

27. Work done in increasing the size of a soap bubble from radius of 3 cm to 5 cm in millijoule is nearly (surface tension of soap solution =  $0.03 \text{ Nm}^{-1}$ )

- (A)  $0.4 \pi$  (B)  $0.2 \pi$  (C)  $4 \pi$  (D)  $2 \pi$

28. What are the values of the currents flowing in each of the following diode circuits X and Y respectively? (Assume that the diodes are ideal)

- (A) 1 A, 2 A  
(B) 2 A, 1 A  
(C) 4 A, 2 A  
(D) 2 A, 4 A



29. In the interference experiment using a biprism, the distance of the slits from the screen is increased by 25% and the separation between the slits is halved. If 'W' represents the original fringewidth, the new fringewidth is

- (A) 2W (B) 2.5 W (C) 4 W (D) 1.5 W

30. Two waves are represented by the equation,  $y_1 = A \sin(\omega t + kx + 0.57) \text{ m}$  and  $y_2 = A \cos(\omega t + kx) \text{ m}$ , where x is in metre and t is in second. What is the phase difference between them?

- (A) 0.57 radian (B) 1.0 radian (C) 1.57 radian (D) 1.25 radian

31. An ideal gas at pressure 'p' is adiabatically compressed so that its density becomes twice that of the initial. If  $\gamma = \frac{c_p}{c_v} = \frac{7}{5}$ , then final pressure of the gas is

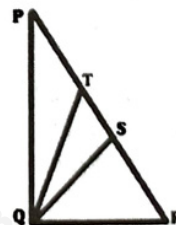
- (A) p (B) 2p (C)  $\frac{7}{5} p$  (D) 2.63p

32. In common emitter amplifier, a change of 0.2 mA in the base current causes a change of 5 mA in the collector current. If input resistance is 2 k $\Omega$  and voltage gain is 75, the load resistance used in the circuit is

- (A) 8 k $\Omega$  (B) 4 k $\Omega$  (C) 12 k $\Omega$  (D) 6 k $\Omega$

33. Figure shows triangular lamina which can rotate about different axes moment of inertia is maximum, about the axis

- (A) PR  
(B) QS  
(C) QR  
(D) PQ





34. Which one of the following statements is wrong for an isobaric process?  
(A) The pressure of the system remains constant  
(B) There is change in volume, when work is done  
(C) Temperature of the system remains constant  
(D) Energy exchanged is used to do work to change internal energy
35. The energy of an electron in the excited hydrogen atom is  $-3.4$  eV. Then according to Bohr's theory, the angular momentum of the electron in that excited state is  
( $h$  = Planks's constant)  
(A)  $\frac{2\pi}{h}$  (B)  $\frac{nh}{2\pi}$  (C)  $\frac{h}{\pi}$  (D)  $\frac{3h}{2\pi}$
36. For a perfectly black body, coefficient of emission is  
(A) zero (B) infinity (C) unity (D) less than one (non-zero)
37. Two rods of different metals have coefficients of linear expansion ' $\alpha_1$ ' and ' $\alpha_2$ ' respectively. Their respective lengths are ' $L_1$ ' and ' $L_2$ '. At all temperatures  $(L_2 - L_1)$  is same. The correct relation is  
(A)  $L_1\alpha_1^2 = L_2\alpha_2^2$  (B)  $L_1^2\alpha_1^2 = L_2^2\alpha_2^2$  (C)  $L_1\alpha_2 = L_2\alpha_1$  (D)  $L_1\alpha_1 = L_2\alpha_2$
38. The temperature of a black body is increased by 50%, then the percentage increase in the rate of radiation by the body is approximated  
(A) 50% (B) 100% (C) 400 % (D) 150%
39. A particle executing S.H.M starts from the mean position. Its amplitude is ' $A$ ' and time period ' $T$ '. At what displacement its speed is one-fourth of the maximum speed?  
(A)  $\frac{A}{\sqrt{15}}$  (B)  $\frac{A}{4}$  (C)  $\frac{4A}{15}$  (D)  $\frac{A\sqrt{15}}{40}$
40. The fundamental frequency of an air column in pipe ' $A$ ' closed at one end coincides with second overtone of pipe ' $B$ ' open at both ends. The ratio of length of pipe ' $A$ ' to that of pipe ' $B$ ' is  
(A) 3:8 (B) 3:4 (C) 1:6 (D) 2:3
41. An electron is projected along the axis of circular conductor carrying current ' $I$ '. The electron will experience  
(A) a force perpendicular to axis (B) a force along the axis  
(C) a force at angle  $30^\circ$  with the axis (D) no force
42. A thin ring of radius ' $R$ ' meter has charge ' $q$ ' coulomb uniformly spread on it. The ring rotates about its axis with a constant frequency of  $f$  revolution/s. The value of magnetic induction in  $\text{Wb m}^{-2}$  at the center of the ring is  
( $\mu_0$  = Permeability of free space)  
(A)  $\frac{\mu_0 q f}{2\pi R}$  (B)  $\frac{\mu_0 q}{2\pi R}$  (C)  $\frac{\mu_0 q f}{2R}$  (D)  $\frac{\mu_0 q}{2\pi f R}$
43. A charged spherical conductor has radius ' $r$ '. The potential difference between its surface and a point at a distance ' $3r$ ' from the centre is ' $v$ '. The electric intensity at a distance ' $3r$ ' from the centre of the conductor is  
(A)  $\frac{v}{8r}$  (B)  $\frac{v}{2r}$  (C)  $\frac{v}{4r}$  (D)  $\frac{v}{6r}$

44. In biprims experiment, the 4<sup>th</sup> dark band is formed opposite to one of the slits. The wavelength of light used is ( $d$  = distance between the slits,  $D$  = distance between source and the screen)

- (A)  $\frac{d^2}{14D}$       (B)  $\frac{d^2}{7D}$       (C)  $\frac{d^2}{9D}$       (D)  $\frac{d^2}{11D}$

45. The magnitude of flux linked with coil varies with time as  $\phi = 3t^2 + 4t + 7$ . The magnitude of induced e.m.f. at  $t = 2$  s is

- (A) 3 V      (B) 16 V      (C) 10 V      (D) 7 V

46. At what rate a single conductor should cut the magnetic flux so that current of 1.5 mA flows through it when a resistance of  $5 \Omega$  is connected across its ends?

- (A)  $6 \times 10^{-3} \frac{\text{wb}}{\text{s}}$       (B)  $8 \times 10^{-3} \frac{\text{wb}}{\text{s}}$   
 (C)  $4 \times 10^{-4} \frac{\text{wb}}{\text{s}}$       (D)  $7.5 \times 10^{-3} \frac{\text{wb}}{\text{s}}$

47. If we increase the frequency of an a.c. supply, then inductive reactance

- (A) increases directly with the square of frequency  
 (B) increases as it directly proportional to frequency  
 (C) decreases inversely with the square of frequency  
 (D) decreases as it is inversely proportional to the frequency

48. An electron of mass 'm' and a photon have same energy 'E'. The ratio of de-Broglie wavelength of electron to the wavelength of photon is ( $c$  = velocity of light)

- (A)  $c\sqrt{\frac{E}{m}}$       (B)  $\frac{1}{c}\sqrt{\frac{2m}{g}}$       (C)  $\frac{1}{c}\sqrt{\frac{E}{2m}}$       (D)  $c\sqrt{\frac{m}{E}}$

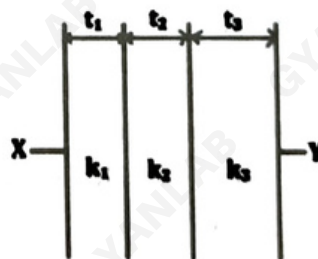
49. Two parallel plates with dielectric placed between the plates are as shown in figure. The resultant capacity of capacitor will be [ $A$  = area of plate.  $t_1$ ,  $t_2$  and  $t_3$  are thickness of dielectric slabs,  $k_1$ ,  $k_2$  and  $k_3$  are dielectric constants.

(A)  $\frac{A\epsilon_0}{\left[ \frac{t_1 + t_2 + t_3}{k_1 + k_2 + k_3} \right]}$

(C)  $A\epsilon_0 \left[ \frac{k_1}{t_1} + \frac{k_2}{t_2} + \frac{k_3}{t_3} \right]$

(B)  $\frac{A\epsilon_0(k_1 k_2 k_3)}{t_1 t_2 t_3}$

(D)  $\frac{A\epsilon_0}{\left[ \frac{t_1}{k_1} + \frac{t_2}{k_2} + \frac{t_3}{k_3} \right]}$



\*50. A bar magnet has length 3 cm, cross-sectional area  $2 \text{ cm}^2$  and magnetic moment  $3 \text{ Am}^2$ . The intensity of magnetisation of bar magnet is

- (A)  $2 \times 10^5 \text{ A/m}$       (B)  $3 \times 10^5 \text{ A/m}$       (C)  $4 \times 10^5 \text{ A/m}$       (D)  $5 \times 10^5 \text{ A/m}$

## CHEMISTRY

51. What is the formal charge on 'N' atom in  $\text{NH}_4^+$  ion?  
 (A) +1 (B) -3 (C) -1 (D) zero
52. Identify most stable free radical from following.  
 (A)  $\text{CH}_3\text{CH}_2\cdot$  (B)  $(\text{CH}_3)_3\text{C}\cdot$  (C)  $(\text{CH}_3)_2\text{CH}\cdot$  (D)  $\text{CH}_3\cdot$
53. Which among the following salt solution in water shows pH less than 7?  
 (A)  $\text{CuCl}_2$  (B)  $\text{CH}_3\text{COONH}_4$  (C)  $\text{Na}_2\text{CO}_3$  (D)  $\text{KNO}_3$
54. Ethoxy benzene on reaction with hot and concentrated HI forms  
 (A) ethyl iodide and phenol (B) ethyl iodide and iodobenzene  
 (C) ethyl alcohol and iodobenzene (D) ethyl alcohol and phenol
55. What is vapour pressure of solution containing 1.8 g glucose in 16.2 g water?  
 ( $P_1^0 = 24$  mm Hg and Molar mass of glucose =  $180 \text{ g mol}^{-1}$ )  
 (A) 18.1 mm Hg (B) 15.7 mm Hg (C) 12.4 mm Hg (D) 23.8 mm Hg
56. The rate law equation for a reaction between A, B and C is  $r = k[\text{A}][\text{B}][\text{C}]^2$ , what will be new rate of reaction if concentration of both A and B are doubled.  
 (A) 2r (B) 4r (C) 6r (D) 8r
57. Identify product C in following conversion.  

$$\text{m-Hydroxy Benzaldehyde} \xrightarrow[\text{[Protection of -OH group]}]{\text{C}_6\text{H}_5\text{CH}_2\text{Cl}} \text{A} \xrightarrow{\text{[O]}} \text{B} \xrightarrow{\text{[deprotection of -OH group]}} \text{C}$$
  
 (A) Benzoic acid (B) m-Hydroxybenzoic acid  
 (C) Phenol (D) Phenyl benzoate
58. Which of the following is NOT dihydric alcohol?  
 (A) Catechol (B) Hydroquinone (C) Phloroglucinol (D) Resorcinol
59. Identify the sugar containing  $\alpha, \beta$ -1,2-glycosidic linkage.  
 (A) Sucrose (B) Maltose (C) Lactose (D) Raffinose
60. What is the freezing point of 1 molal aqueous solution of a non volatile solute?  
 ( $K_f = 1.86 \text{ K kg mol}^{-1}$ ) ( $T_f^0$  for water =  $0^\circ\text{C}$ )  
 (A)  $-0.93^\circ\text{C}$  (B)  $-2.43^\circ\text{C}$  (C)  $-3.72^\circ\text{C}$  (D)  $-1.86^\circ\text{C}$
61. How many particles per unit cell are present in BCC structure?  
 (A) 1 (B) 4 (C) 3 (D) 2
62. Identify ferromagnetic element from following.  
 (A) Iron (B) Vanadium (C) Chromium (D) Manganese
63. Identify the reagent used in following conversion.  

$$\text{Pent-3-enitrile} \xrightarrow{\text{A}} \text{pent-3-enal}$$
  
 (A)  $\text{AlH}(\text{i-Bu})_2/\text{H}_3\text{O}^+$  (B)  $\text{K}_2\text{Cr}_2\text{O}_7/\text{H}_2\text{SO}_4$   
 (C)  $\text{H}_2/\text{Pd.BaSO}_4$  (D)  $\text{SnCl}_2 \cdot \text{HCl}$



64. Which of the following complexes is diamagnetic and square planar?  
 (A)  $[\text{CoF}_6]^{3-}$  (B)  $[\text{Co}(\text{NH}_3)_6]^{3+}$  (C)  $[\text{NiCl}_4]^{2-}$  (D)  $[\text{Ni}(\text{CN})_4]^{2-}$
65. What is the mass of  $33.6 \text{ dm}^3$  of methane gas at S.T.P.?  
 (A)  $4.8 \times 10^{-2} \text{ kg}$  (B)  $3.3 \times 10^{-2} \text{ kg}$   
 (C)  $1.6 \times 10^{-2} \text{ kg}$  (D)  $2.4 \times 10^{-2} \text{ kg}$
66. An element with BCC structure has edge length of 500 pm. If its density is  $4 \text{ g cm}^{-3}$ , find atomic mass of the element?  
 (A)  $150 \text{ g mol}^{-1}$  (B)  $100 \text{ g mol}^{-1}$  (C)  $125 \text{ g mol}^{-1}$  (D)  $250 \text{ g mol}^{-1}$
67. Identify the reagent (A) used in the following reaction.
- $$\text{C}_2\text{H}_5\text{MgBr} \xrightarrow{\text{A}} \text{C}_2\text{H}_6 + \text{Mg} \begin{array}{l} \text{Br} \\ \diagup \\ \text{NH}_2 \end{array}$$
- (A)  $\text{NH}_3$  (B)  $\text{RNH}_2$  (C)  $\text{H}_2\text{O}$  (D)  $\text{CH}_3\text{OH}$
68. What is the product formed when side chain chlorination of toluene is carried out followed by acid hydrolysis at 373 K?  
 (A) Benzaldehyde (B) Benzal chloride  
 (C) Chlorobenzene (D) Benzoic acid
69. A system gives out  $x \text{ J}$  of heat and does  $y \text{ J}$  of work on its surrounding. What is the internal energy change?  
 (A)  $-x - y \text{ J}$  (B)  $y - x \text{ J}$  (C)  $x - y \text{ J}$  (D)  $x + y \text{ J}$
70. Which of following reactions yields biphenyl from chlorobenzene?  
 (A) Swartz reaction (B) Wurtz reaction  
 (C) Fittig reaction (D) Finkelstein reaction
71. A system does 394 J of work on surrounding by absorbing 701 J heat. What is the change in internal energy of the system?  
 (A) 547 J (B) 1095 J (C) 307 J (D) 394 J
72. How many electrons flow through the wire if a current of 1.5 ampere flow through it for 3 hours?  
 (A)  $1.60 \times 10^{19}$  (B)  $1.01 \times 10^{23}$  (C)  $1.01 \times 10^{19}$  (D)  $1.60 \times 10^{23}$
73. The conductivity of 0.3 M solution of KCl at 298 K is  $0.0627 \text{ S cm}^{-1}$ . What is its molar conductivity?  
 (A)  $104 \text{ S cm}^2 \text{ mol}^{-1}$  (B)  $188 \text{ S cm}^2 \text{ mol}^{-1}$   
 (C)  $209 \text{ S cm}^2 \text{ mol}^{-1}$  (D)  $109 \text{ S cm}^2 \text{ mol}^{-1}$
74. The solubility of  $\text{Ag}_2\text{C}_2\text{O}_4$  is  $2 \times 10^{-4} \text{ mol L}^{-1}$  at 298 K. What is its solubility product?  
 (A)  $1.6 \times 10^{-6}$  (B)  $3.2 \times 10^{-11}$  (C)  $1.6 \times 10^{-11}$  (D)  $3.2 \times 10^{-6}$
75. Two electrons occupying the same orbital are distinguished by  
 (A) Principal quantum number (B) Azimuthal quantum number  
 (C) Magnetic quantum number (D) Spin quantum number



76. Which statement from following is correct for homolytic fission?  
(A) In this single electron of shared pair moves to one of bonded atom and other to other atom.  
(B) In this electron deficient and electron rich species are formed.  
(C) In this electronegative atom pulls both electrons of shared pair.  
(D) In this carbon free radical formed has  $sp^3$  hybridisation.
77. Identify catalyst used in manufacturing of HDP.  
(A) Ziegler – Natta (B) Peroxides  
(C) Lindlar's (D) Magnesium oxide
78. Air is an example of a solution of  
(A) gas in solid (B) liquid in gas (C) gas in liquid (D) gas in gas
79. A gas has a volume of 3.4 L at 298 K. What is the final temperature if the volume of gas increases to 6.8 L?  
(A) 596 K (B) 412 K (C) 298 K (D) 149 K
80. What is the atomic radius of polonium if it crystallises in a simple cubic structure with edge length of unit cell 336 pm?  
(A) 84 pm (B) 168 pm (C) 234 pm (D) 336 pm
81. Which among the following halides has trigonal bipyramidal structure?  
(A)  $SeCl_2$  (B)  $SeF_4$  (C)  $SF_6$  (D)  $TeF_6$
82. Which among following compounds is used as monomer in preparation of Teflon?  
(A) Tetrabromoethylene (B) Tetrafluoroethylene  
(C) Tetrachloroethylene (D) Tetraiodoethylene
83. Which of the following is NOT obtained when a mixture of bomomethane and bromoethane is treated with sodium in dry ether?  
(A) Propane (B) Butane (C) Methane (D) Ethane
84. Time required for 90% completion of a first order reaction is t. What is the time required for completion of 99% reaction?  
(A) t (B) 2t (C)  $t/2$  (D) 3t
85. A weak monobasic acid is 3.0% dissociated in it's 0.04 M solution. What is the dissociation constant of acid?  
(A)  $9 \times 10^{-4}$  (B)  $3.6 \times 10^{-5}$  (C)  $3 \times 10^{-2}$  (D)  $4 \times 10^{-2}$
86. Which of the following alkanes is optically active?  
(A) 2-Methylbutane (B) 2,3-Dimethylbutane  
(C) 2,3-Dimethylpentane (D) 2-Methylpropane
87. Identify the number of unpaired electrons present and geometry respectively of  $[Co(NH_3)_6]^{3+}$  complex.  
(A) 0, square planar (B) 2, square planar (C) 4, octahedral (D) 0, octahedral
88. Which among the following compounds is a weakest base?  
(A) Phenymethanamine (B) N-Methylaniline  
(C) Benzenamine (D) N,N-Dimethylaniline

89. In the cell represented as  $\text{Ni}_{(s)} \mid \text{Ni}_{(IM)}^{2\oplus} \parallel \text{Ag}_{(IM)}^{\oplus} \mid \text{Ag}_{(s)}$ , the reducing agent is

- (A) Ag                      (B)  $\text{Ag}^{\oplus}$                       (C) Ni                      (D)  $\text{Ni}^{2\oplus}$

90. What is the value of intercept on y-axis when  $\log \frac{x}{m}$  is plotted against  $\log P$  in Freundlich isotherm?

- (A)  $\frac{1}{n}$                       (B)  $\log k$                       (C)  $n$                       (D)  $k$

91. Which from following statements is true for group 16 elements?

- (A) All elements of this group form  $\text{EO}_2$  type oxides.  
 (B) It includes all the nonmetals.  
 (C) Oxides of all elements of this group are gaseous at room temperature.  
 (D) Reducing properties of dioxides of this group element decreases from  $\text{SO}_2$  to  $\text{TeO}_2$ .

92. Which among the following formulae is correctly represented according to stock notation?

- (A)  $\text{Fe(II)Cl}_3$                       (B)  $\text{Mn(II)O}_2$                       (C)  $\text{Au(III)Cl}$                       (D)  $\text{Sn(IV)Cl}_4$

93. Identify the major product when anisole is treated with  $\text{Br}_2$  in presence of acetic acid.

- (A) m-Bromo anisole                      (B) p-Bromo anisole  
 (C) o-Bromo anisole                      (D) 2,4,6-Tribromo anisole

94. All the elements of group 2 react with water to form metal hydroxide and hydrogen, except the element

- (A) Barium                      (B) Calcium                      (C) Magnesium                      (D) Beryllium

95. Which of the following compound is obtained when glucose is treated with dilute nitric acid?

- (A) Glucose oxime                      (B) Gluconic acid  
 (C) Saccharic acid                      (D) Glucose cyanohydrin

96. What is the IUPAC name of glyoxal?

- (A) Propanedial                      (B) Ethanedial  
 (C) 2-Methyl propanal                      (D) Prop-2-enal

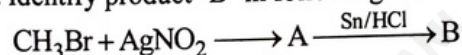
97. The order of reaction for which the units of rate constant are  $\text{mol dm}^{-3} \text{s}^{-1}$  is

- (A) 2                      (B) 1                      (C) 0                      (D) 3

98. Which among the following is a nonferrous alloy?

- (A) Brass                      (B) Nickel steel                      (C) Stainless steel                      (D) Chromium steel

99. Identify product 'B' in following reaction.



- (A)  $\text{CH}_3\text{NO}_2$                       (B)  $\text{CH}_3\text{NH}_2$                       (C)  $\text{CH}_3\text{Cl}$                       (D)  $\text{CH}_3\text{OH}$

\*100. A balloon contains 2.27 L air and has a pressure of  $1.013 \times 10^5 \text{ Nm}^{-2}$ . The balloon rises to a certain height and expands to volume of 4540 mL. What is the final pressure of air in balloon?

- (A)  $2.026 \times 10^2 \text{ Nm}^{-2}$                       (B)  $5.065 \times 10^4 \text{ Nm}^{-2}$   
 (C)  $4.540 \times 10^4 \text{ Nm}^{-2}$                       (D)  $5.065 \times 10^{-4} \text{ Nm}^{-2}$



## Section II

## MATHEMATICS

101. If  $A(3, 2, -1)$ ,  $B(-2, 2, -3)$  and  $D(-2, 5, -4)$  are the vertices of a parallelogram, then the area of the parallelogram is

- (A) 286 sq. units (B)  $\sqrt{286}$  sq. units  
(C) 300 sq. units (D)  $\sqrt{300}$  sq. units

102. If  $\hat{a}$  is a unit vector such that  $(\bar{x} - \hat{a}) \cdot (\bar{x} + \hat{a}) = 8$ , then  $|\bar{x}| =$

- (A)  $\pm 3$  (B)  $2\sqrt{2}$  (C) 3 (D)  $\pm\sqrt{7}$

103. The equation of line, where length of the perpendicular segment from origin to the line is 4 and the inclination of this perpendicular segment with the positive direction of X-axis is  $30^\circ$ , is

- (A)  $x + \sqrt{3}y = 8$  (B)  $x - \sqrt{3}y = 8$   
(C)  $\sqrt{3}x - y = 8$  (D)  $\sqrt{3}x + y = 8$

104. If  $\int \frac{\sin x}{\sin(x - \alpha)} dx = Ax + B \log \sin(x - \alpha) + c$ , then the value of A and B are respectively

- (where c is a constant of integration)  
(A)  $\cos \alpha, \sin \alpha$  (B)  $\sin \alpha, \cos \alpha$   
(C)  $-\cos \alpha, \sin \alpha$  (D)  $-\sin \alpha, \cos \alpha$

105. The probability distribution of the number of doublets in four throws of a pair of dice is given by

(A)

|              |               |               |               |               |               |
|--------------|---------------|---------------|---------------|---------------|---------------|
| <b>X:</b>    | 0             | 1             | 2             | 3             | 4             |
| <b>P(X):</b> | $\frac{1}{5}$ | $\frac{1}{5}$ | $\frac{1}{5}$ | $\frac{1}{5}$ | $\frac{1}{5}$ |

(B)

|              |               |               |               |               |
|--------------|---------------|---------------|---------------|---------------|
| <b>X:</b>    | 0             | 1             | 2             | 3             |
| <b>P(X):</b> | $\frac{1}{4}$ | $\frac{1}{4}$ | $\frac{1}{4}$ | $\frac{1}{4}$ |

(C)

|              |               |               |               |               |
|--------------|---------------|---------------|---------------|---------------|
| <b>X:</b>    | 1             | 2             | 3             | 4             |
| <b>P(X):</b> | $\frac{1}{4}$ | $\frac{1}{4}$ | $\frac{1}{4}$ | $\frac{1}{4}$ |

(D)

|              |                    |                   |                  |                 |                  |
|--------------|--------------------|-------------------|------------------|-----------------|------------------|
| <b>X:</b>    | 0                  | 1                 | 2                | 3               | 4                |
| <b>P(X):</b> | $\frac{625}{1296}$ | $\frac{125}{324}$ | $\frac{25}{216}$ | $\frac{5}{324}$ | $\frac{1}{1296}$ |

106. Let  $\vec{v} = 2\hat{i} + 2\hat{j} - \hat{k}$  and  $\vec{w} = \hat{i} + 3\hat{k}$ . If  $\vec{u}$  is a unit vector, then the maximum value of the scalar triple product  $[\vec{u} \vec{v} \vec{w}]$  is

- (A)  $\sqrt{6}$  (B)  $\sqrt{10}$  (C)  $\sqrt{13}$  (D)  $\sqrt{89}$

107. S1 : If  $-7$  is an integer, then  $\sqrt{-7}$  is a complex number

S2 :  $-7$  is not an integer or  $\sqrt{-7}$  is a complex number

- (A) S1 and S2 are converse statements of each other  
(B) S1 and S2 are negations of each other  
(C) S1 and S2 are equivalent statements  
(D) S1 and S2 are contrapositive of each other



108. For the probability distribution given by following

|               |             |            |            |          |             |             |             |
|---------------|-------------|------------|------------|----------|-------------|-------------|-------------|
| <b>x</b>      | <b>5</b>    | <b>6</b>   | <b>7</b>   | <b>8</b> | <b>9</b>    | <b>10</b>   | <b>11</b>   |
| <b>P(X=x)</b> | <b>0.07</b> | <b>0.2</b> | <b>0.3</b> | <b>k</b> | <b>0.07</b> | <b>0.04</b> | <b>0.02</b> |

$$\text{Var}(X) =$$

- (A) 2.65                      (B) 2.85                      (C) 1.65                      (D) 3.85

$$109. \int_0^1 |5x - 3| dx =$$

- (A)  $\frac{13}{10}$                       (B) 1                      (C)  $\frac{3}{10}$                       (D)  $\frac{1}{2}$

110. Function  $f(x) = e^{-1/x}$  is strictly increasing for all x where

- (A) x is only positive real number                      (B) x is only negative real number  
(C) x is a real number                      (D) x is a non-zero real number

111. The distance between the parallel lines

$$\frac{x-2}{3} = \frac{y-4}{5} = \frac{z-1}{2} \text{ and } \frac{x-1}{3} = \frac{y+2}{5} = \frac{z+3}{2} \text{ is}$$

- (A)  $\frac{1}{\sqrt{38}}$  units                      (B)  $\sqrt{\frac{333}{38}}$  units                      (C)  $\sqrt{\frac{300}{37}}$  units                      (D)  $\sqrt{\frac{300}{35}}$  units

112. The general solution of the differential equation

$$\frac{dy}{dx} + \frac{y^2 + y + 1}{x^2 + x + 1} = 0 \text{ is}$$

- (A)  $x + y + 1 = c(1 + x + y + 2xy)$                       (B)  $x + y + 1 = c(2 + x + y + 2xy)$   
(C)  $x + y + 1 = c(1 - x - y - 2xy)$                       (D)  $x + y + 2 = c(2 - x - y - 2xy)$

$$113. \tan\left(\cos^{-1}\left(\frac{4}{5}\right) + \tan^{-1}\left(\frac{2}{3}\right)\right) =$$

- (A)  $\frac{17}{6}$                       (B)  $\frac{17}{3}$                       (C)  $\frac{18}{5}$                       (D)  $\frac{7}{15}$

114. The general solution of the differential equation  $\frac{dx}{dt} = \frac{x \log x}{t}$  is

- (A)  $\log x - x = c$                       (B)  $e^{ct} + x = 0$   
(C)  $\log t = x + c$                       (D)  $e^{ct} = x$

115. If  $3\sin\theta = 2\sin 3\theta$  and  $0 < \theta < \pi$ , then  $\sin\theta =$

- (A)  $\frac{\sqrt{2}}{\sqrt{5}}$                       (B)  $\frac{\sqrt{3}}{2\sqrt{2}}$                       (C)  $\frac{\sqrt{2}}{3}$                       (D)  $\frac{\sqrt{3}}{\sqrt{5}}$

116. The maximum value of the objective function  $z = 2x + 3y$  subject to the constraints

$$x + y \leq 5, 2x + y \geq 4 \text{ and } x \geq 0, y \geq 0 \text{ is}$$

- (A) 15                      (B) 10                      (C) 20                      (D) 25

117. The particular solution of differential equation

$$(x + y)dy + (x - y)dx = 0 \text{ at } x = y = 1 \text{ is}$$

- (A)  $\log \left| \frac{x^2 + y^2}{2} \right| = \frac{\pi}{2} - 2 \tan^{-1} \left( \frac{y}{x} \right)$       (B)  $\log |x^2 + y^2| = \frac{\pi}{2} - 2 \tan^{-1} \left( \frac{y}{x} \right)$   
(C)  $\log |x^2 + y^2| = \frac{\pi}{2} - 2 \tan^{-1} \left( \frac{y}{x} \right)$       (D)  $\log \left| \frac{x^2 + y^2}{2} \right| = \frac{\pi}{4} - 2 \tan^{-1} \left( \frac{y}{x} \right)$

118.  $\int \frac{10^{\frac{x}{2}}}{\sqrt{10^{-x} - 10^x}} dx =$

- (A)  $2\sqrt{10^{-x} + 10^x} + c$       (B)  $\frac{2}{2\sqrt{10^{-x} + 10^x}} + c$   
(C)  $\frac{1}{\log 10} \sin^{-1}(10^x) + c$       (D)  $\frac{1}{\log 10} \cos^{-1}(10^x) + c$

119.  $\int_0^{\pi/2} \frac{\cos x}{3 \cos x + \sin x} dx =$

- (A)  $\frac{3\pi}{20} - \frac{\log 3}{100}$       (B)  $\frac{3\pi}{10} - \frac{\log 3}{10}$       (C)  $\frac{3\pi}{20} + \frac{\log 3}{10}$       (D)  $\frac{3\pi}{20} - \frac{\log 3}{10}$

120. If  $A = \begin{bmatrix} 2 & 3 \\ 5 & -2 \end{bmatrix}$  and  $A^{-1} = KA$ , then K is

- (A) 19      (B)  $-\frac{1}{19}$       (C) -19      (D)  $\frac{1}{19}$

121. If  $A = \begin{bmatrix} 1 & 2 & 3 \\ -1 & 1 & 2 \\ 1 & 2 & 4 \end{bmatrix}$ , then  $A(\text{adj } A) =$

- (A)  $\begin{bmatrix} -1/3 & 0 & 0 \\ 0 & -1/3 & 0 \\ 0 & & -1/3 \end{bmatrix}$       (B)  $\begin{bmatrix} 3 & 0 & 0 \\ 0 & 3 & 0 \\ 0 & 0 & 3 \end{bmatrix}$   
(C)  $\begin{bmatrix} 1 & 2 & 3 \\ -1 & 1 & 2 \\ 1 & 2 & 4 \end{bmatrix}$       (D)  $\begin{bmatrix} 1 & -1 & 1 \\ 2 & 1 & 2 \\ 3 & 2 & 4 \end{bmatrix}$

122. A random variable X has the following probability distribution

| x      | 0 | 1  | 2  | 3  | 4  | 5  | 6  | 7 | 8 |
|--------|---|----|----|----|----|----|----|---|---|
| P(X=x) | K | 2K | 3K | 4K | 4K | 3K | 2K | K | K |

Then  $P(3 < x \leq 6) =$

- (A)  $\frac{3}{7}$       (B)  $\frac{4}{7}$       (C)  $\frac{13}{21}$       (D)  $\frac{8}{21}$

123. If  $x = -2$  and  $x = 4$  are the extreme points of  $y = x^3 - \alpha x^2 - \beta x + 5$ , then

- (A)  $\alpha = 3, \beta = 24$  (B)  $\alpha = -24, \beta = -3$   
 (C)  $\alpha = -3, \beta = -24$  (D)  $\alpha = 24, \beta = 3$

124. The coordinates of the foot of the perpendicular drawn from the origin to the plane  $2x + y - 2z = 18$  are

- (A) (4, 2, -4) (B) (1, 2, -3) (C) (4, 2, 4) (D) (4, -2, -4)

125. If a circle passes through the points (0, 0), (x, 0) and (0, y), then the coordinates of its centre are

- (A)  $\left(\frac{-x}{2}, \frac{y}{2}\right)$  (B)  $\left(\frac{x}{2}, \frac{y}{2}\right)$  (C)  $\left(\frac{-x}{2}, \frac{-y}{2}\right)$  (D)  $\left(\frac{x}{2}, \frac{-y}{2}\right)$

126. If  $A = \begin{bmatrix} 1 & -1 & 1 \\ 2 & -1 & 0 \\ 3 & 3 & -4 \end{bmatrix}$ ,  $B = \begin{bmatrix} 1 \\ 1 \\ 2 \end{bmatrix}$  and  $X = \begin{bmatrix} x_1 \\ x_2 \\ x_3 \end{bmatrix}$  such that  $AX = B$ , then the value of  $x_1 + x_2 + x_3 =$

- (A) 4 (B) 5 (C) 6 (D) 3

127. If  $\vec{a} = \hat{i} + 2\hat{j} + 3\hat{k}$ ,  $\vec{b} = -\hat{i} + 2\hat{j} + \hat{k}$ ,  $\vec{c} = 3\hat{i} + \hat{j}$  and  $\vec{a} + \lambda\vec{b}$  is perpendicular to  $\vec{c}$ , then  $\lambda =$

- (A) 5 (B) 2 (C) 3 (D) 4

128. If two lines represented by  $ax^2 + 2hxy + by^2 = 0$  makes angles  $\alpha$  and  $\beta$  with positive direction of X-axis, then  $\tan(\alpha + \beta) =$

- (A)  $\frac{2h}{b-a}$  (B)  $\frac{2h}{a-b}$  (C)  $\frac{h}{a+b}$  (D)  $\frac{2h}{a+b}$

129. If  $\tan^{-1} \left[ \frac{\sqrt{1+x^2} - \sqrt{1-x^2}}{\sqrt{1+x^2} + \sqrt{1-x^2}} \right] = \alpha$ , then the value of  $\sin 2\alpha$  is

- (A)  $x^3$  (B)  $\sqrt{x}$  (C)  $x$  (D)  $x^2$

130. The general solution of the differential equation  $\frac{dy}{dx} = 2^{y-x}$  is

- (A)  $2^x - 2^y = c$  (B)  $\frac{1}{2^x} - \frac{1}{2^y} = c$   
 (C)  $\frac{1}{2^x} + \frac{1}{2^y} = c$  (D)  $2^x + 2^y = c$

131. The vector equation of the line passing through P(1, 2, 3) and Q(2, 3, 4) is

- (A)  $(\hat{i} + 2\hat{j} + 3\hat{k}) + \lambda(\hat{i} + \hat{j} + \hat{k})$  (B)  $(\hat{i} + 2\hat{j} + 3\hat{k}) + \lambda(\hat{i} - \hat{j} - \hat{k})$   
 (C)  $(\hat{i} + 2\hat{j} + 3\hat{k}) + \lambda(2\hat{i} + 3\hat{j} + 4\hat{k})$  (D)  $(\hat{i} + 2\hat{j} + 3\hat{k}) + \lambda(2\hat{i} + 6\hat{j} + 12\hat{k})$

132. Range of the function  $f(x) = 3 + 2^x + 4^x$  is

- (A) (3,  $\infty$ ) (B)  $(-\infty, \infty)$  (C) [3,  $\infty$ ) (D)  $(-\infty, 3]$



133. The combined equation of a pair of lines passing through the origin and inclined at  $60^\circ$  and  $30^\circ$  respectively with x-axis is

- (A)  $\sqrt{3}(x^2 + y^2) = 2xy$  (B)  $\sqrt{3}(x^2 + y^2) = 4xy$   
(C)  $4(x^2 + y^2) = \sqrt{3}xy$  (D)  $2(x^2 + y^2) = \sqrt{3}xy$

134.  $\int e^{(e^x+x)} dx =$

- (A)  $e^x + x + c$  (B)  $e^{(e^x)} \cdot x + c$   
(C)  $e^{(e^x)} + c$  (D)  $e^{(e^x)}(e^x - 1) + c$

135. If  $z = x + iy$  satisfies the condition  $|z + 1| = 1$ , then  $z$  lies on the

- (A) parabola with vertex  $(0, 0)$   
(B) circle with centre  $(-1, 0)$  and radius 1  
(C) circle with centre  $(1, 0)$  and radius 1  
(D) Y-axis

136. If  $3\hat{j}, 4\hat{k}$  and  $3\hat{j} + 4\hat{k}$  are the position vectors of the vertices A, B, C respectively of  $\triangle ABC$ , then the position vector of the point in which the bisector of  $\angle A$  meets BC is

- (A)  $\frac{5}{3}\hat{j} - 4\hat{k}$  (B)  $5\hat{j} - 4\hat{k}$  (C)  $5\hat{j} + 4\hat{k}$  (D)  $\frac{5}{3}\hat{j} + 4\hat{k}$

137. 10 is divided into two parts such that the sum of double of the first and square of the other is minimum, then the numbers are respectively

- (A) 9, 1 (B) 8, 2 (C) 6, 4 (D) 7, 3

138. In a triangle ABC, with usual notations  $a = 2, b = 3, c = 5$ , then  $\frac{\cos A}{a} + \frac{\cos B}{b} + \frac{\cos C}{c} =$

- (A)  $\frac{19}{30}$  (B)  $\frac{19}{60}$  (C)  $\frac{23}{60}$  (D)  $\frac{38}{35}$

139. If  $f(x) = \frac{1 - \sin x + \cos x}{1 + \sin x + \cos x}$ , for  $x \neq \pi$  is continuous at  $x = \pi$ , then the value of  $f(\pi)$  is

- (A)  $-\frac{1}{2}$  (B)  $-1$  (C)  $1$  (D)  $\frac{1}{2}$

140. Given that total of 16 values is 528 and sum of the squares of deviation from 33 is 9158. The variance is

- (A) 562.73 (B) 570.375 (C) 574.375 (D) 572.375

141.  $\lim_{x \rightarrow 1} \left[ \frac{\sqrt{x} - 1}{\log x} \right] =$

- (A)  $\frac{1}{2}$  (B) 2 (C)  $-2$  (D)  $-\frac{1}{2}$

142. If the surrounding air is kept at 25°C and a body cools from 80°C to 50°C in 30 minutes, then temperature of the body after one hour will be

- (A) 31.72°C approximately (B) 34.74°C approximately  
(C) 32.36°C approximately (D) 36.36°C approximately

143. Rooms in a hotel are numbered from 1 to 19. Rooms are allocated at random as guests arrive. The first guest to arrive is given a room which is a prime number. The probability that the second guest to arrive is given a room which is a prime number is

- (A)  $\frac{8}{19} \times \frac{7}{18}$  (B)  $\frac{8}{19}$  (C)  $\frac{8}{19} \times \frac{7}{19}$  (D)  $\frac{7}{18}$

144. Negation of the statement :  $3 + 6 > 8$  and  $2 + 3 < 6$  is

- (A)  $3 + 6 \leq 8$  or  $2 + 3 < 6$  (B)  $3 + 6 < 8$  or  $2 + 3 < 6$   
(C)  $3 + 6 \leq 8$  or  $2 + 3 \geq 6$  (D)  $3 + 6 > 8$  or  $2 + 3 \geq 6$

145. If  $f(x) = \operatorname{cosec}^{-1} \left[ \frac{10}{6 \sin(2^x) - 8 \cos(2^x)} \right]$ , then  $f'(x) =$

- (A)  $2^x \log 2$  (B)  $-1$  (C)  $\log 2$  (D)  $2^x$

146. The area bounded by the parabola  $y^2 = 4ax$  and its latus-rectum  $x = a$  is

- (A)  $\frac{8}{3} a^2$  sq. units (B)  $\frac{2}{3} a^2$  sq. units  
(C)  $\frac{4}{3} a^2$  sq. units (D)  $8 a^2$  sq. units

147. If  $y = \log \sqrt{\tan x}$ , then the value of  $\frac{dy}{dx}$  at  $x = \frac{\pi}{4}$  is

- (A) 1 (B)  $-1$  (C)  $\frac{1}{2}$  (D) 0

148. If  $x = a \left( t - \frac{1}{t} \right)$  and  $y = b \left( t + \frac{1}{t} \right)$ , then  $\frac{dy}{dx} =$

- (A)  $\frac{a^2 x}{b^2 y}$  (B)  $\frac{a^2 y}{b^2 x}$  (C)  $\frac{-b^2 x}{a^2 y}$  (D)  $\frac{b^2 x}{a^2 y}$

149. Out of 7 consonants and 4 vowels, the number of words consisting of 3 consonants and 2 vowels are

- (A) 3300 (B) 210 (C) 120 (D) 25200

\*150. Equation of planes parallel to the plane  $x - 2y + 2z + 4 = 0$  which are at a distance of one unit from the point (1, 2, 3) are

- (A)  $x + 2y + 2z = 6$ ,  $x + 2y + 2z = 0$  (B)  $x - 2y + 2z = 0$ ,  $x - 2y + 2z - 6 = 0$   
(C)  $x - 2y - 6 = 0$ ,  $x - 2y + 2z = 6$  (D)  $x + 2y + 2z = -6$ ,  $x + 2y + 2z = 5$